
 Communications to the Editor

 ISOLATION AND STRUCTURAL
 ELUCIDATION OF ANTIOXIDATIVE
 AGENTS, ANTIOSTATINS A₁
 TO A₄ AND B₂ TO B₅

Sir:

Active oxygen species cause a variety of diseases such as ischemia-reperfusion, inflammation, auto-immune disease, diabetes, rheumatism, cardiovascular diseases and cancer-initiation^{1,2}. Thus, it could be expected that antioxidative agents may prevent these diseases.

During the course of a screening program for novel compounds showing antioxidant activity, we obtained a new naphthoquinone derivative naphterpin³. Further screening resulted in the isolation of new antioxidative agents named antiostatins A₁ to A₄ and B₂ to B₅ from *Streptomyces cyaneus* 2007-SV₁. They showed strong inhibitory activity against lipid peroxidation induced by free radicals in rat liver microsomes free from vitamin E⁴. In this paper, we report the fermentation, isolation and structural studies of these new metabolites.

A stock culture of *S. cyaneus* 2007-SV₁ was inoculated into 500-ml Erlenmeyer flasks containing 100 ml of a seed medium (starch 2.5%, soybean meal 1.5%, dry yeast 0.2% and CaCO₃ 0.4%, pH 6.2 before sterilization) and incubated at 27°C for 2 days on a rotary shaker. The seed culture (600 ml) thus obtained was transferred into a 50-liter jar fermenter containing 30 liters of the same medium, and cultivation was carried out at 27°C for 48 hours under agitation at 400 rpm and aeration at 30 liters per minute.

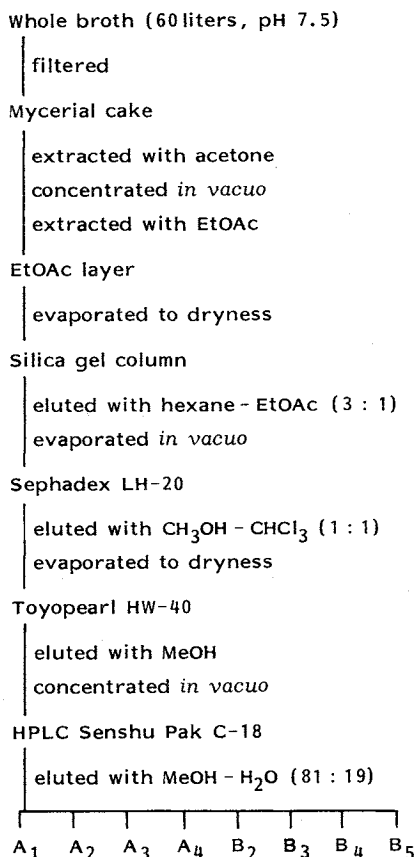
The active materials were isolated according to the procedures shown in Scheme 1. Each component of the antiostatins was separated at the final stage by reversed phase HPLC (Senshu Pak C-18) and was obtained as a pale yellow powder. The antiostatins were soluble in MeOH, CHCl₃, acetone and ethyl acetate, but insoluble in hexane and water. Based on the structural features of the substituent at C-4, they were divided into two groups, the A series and B series. Their structural studies were carried out using the main components A₁ (yield 2.5 mg) and B₄ (yield 2.0 mg) as explained in the following.

Antiostatin A₁: MP 180~183°C (dec); UV $\lambda_{\max}^{\text{MeOH}}$ nm (ϵ) 220 (28,500), 238 (26,200), 301 (16,100), 338 (4,250), 350 (4,250); IR ν_{\max} (KBr) cm⁻¹ 3350, 1620, 1585; HREI-MS (m/z) 324.1822 (M⁺), calcd for C₂₀H₂₄N₂O₂, 324.1838.

Antiostatin B₄: MP 118~120°C[†]; UV $\lambda_{\max}^{\text{MeOH}}$ nm (ϵ) 218 (44,300), 238 (35,300), 301 (20,900), 338 (6,160), 352 (6,290); IR ν_{\max} (KBr) cm⁻¹ 3480, 3370, 3340, 1690, 1670, 1550; HRFAB-MS (m/z) 452.2763 (M⁺), calcd for C₂₆H₃₆N₄O₃, 452.2787.

The UV spectral similarity between antiostatins A₁ and B₄, and carbazomycin B^{5,6} suggested the presence of the carbazole nucleus in these compounds. Their structures were determined by ¹H and ¹³C NMR spectral analyses and by comparison

Scheme 1. Isolation and purification of antiostatins.



[†] Antiostatins B₂ to B₅ gradually started to sublime at around 90°C.

Table 1. ^{13}C NMR data of antiostatins A_1 and B_4 , and carazostatin^a.

Carbon	A_1	B_4	Carazostatin
1	123.3 (s)	123.7 (s)	124.1 (s) ^b
2	125.8 (s)	126.0 (s)	121.4 (s)
3	144.3 (s)	143.9 (s)	148.2 (s)
4	118.3 (s)	117.6 (s)	103.0 (s)
4a	115.5 (s)	115.7 (s)	120.9 (s)
4b	123.8 (s)	123.7 (s)	123.7 (s)
5	122.8 (d)	122.6 (d)	120.0 (d)
6	119.2 (d)	119.7 (d)	118.9 (d)
7	125.8 (d)	126.0 (d)	125.2 (d)
8	111.8 (d)	112.1 (d)	110.6 (d)
8a	141.4 (s)	141.8 (s)	139.8 (s)
9a	135.1 (s)	135.3 (s)	134.0 (s)
10	14.6 (q)	13.3 (q)	12.0 (q)
1'	30.0 (t)	29.7 (t)	28.8 (t)
2'	30.1 (t)	31.1 (t)	29.5 (t)
3'	33.1 (t)	31.0 (t)	30.0 (t)
4'	23.5 (t)	30.6 (t)	29.3 (t)
5'	13.0 (q)	33.2 (t)	31.7 (t)
6'	—	23.8 (t)	22.7 (t)
7'	—	14.9 (q)	14.1 (q)
1''	172.1 (s)	155.9 (s)	—
2''	23.7 (q)	156.8 (s)	—
3''	—	48.3 (t)	—
4''	—	30.1 (d)	—
5''	—	20.7 (q)	—
6''	—	20.7 (q)	—

^a Solvent: A_1 and B_4 in $(\text{CD}_3)_2\text{CO}$, carazostatin in CDCl_3 .

^b q = CH_3 , t = CH_2 , d = CH, s = -C.

with a carbazomycin related compound carazostatin⁷⁾ (Fig. 2).

Detailed analysis of the ^1H NMR spectral data (500 MHz, acetone- d_6) of antiostatin A_1 revealed the presence of an *n*-pentyl side chain (CH_3 , t; 0.89 ppm; $3 \times \text{CH}_2$, each m, 1.37, 1.45 and 1.65 ppm; CH_2 , t, 2.97 ppm), an *N*-acetyl group (CH_3 , s, 2.47 ppm, NH, br s, 9.68 ppm), an aromatic methyl group (CH_3 , s, 2.40 ppm), a 1,2-disubstituted benzene ring (8.13, 7.10, 7.30 and 7.45 ppm), a phenolic hydroxy group (s, 8.05 ppm) and an imino proton of a carbazole nucleus (br s, 10.16 ppm). Based on heteronuclear multiple-bond correlation (HMBC)⁸⁾ spectral analysis (*vide infra*) and comparison of the ^{13}C NMR spectral data (125 MHz, acetone- d_6) of antiostatin A_1 and carazostatin (Table 1), these groups were arranged on the carbazole nucleus as shown in Fig. 1.

Thus, the methylene proton (1'-H, 2.97 ppm) of the *n*-pentyl side chain was coupled to C-9a (135.1 ppm), C-1 (123.3 ppm) and C-2 (125.8 ppm). The aromatic methyl proton was coupled to the last

Fig. 1. Partial structure of antiostatin A_1 as revealed by HMBC analysis.

Arrows indicate ^{13}C - ^1H long range couplings.

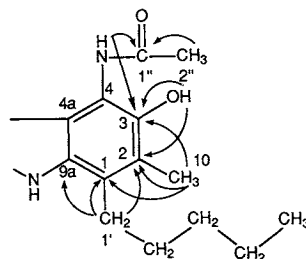
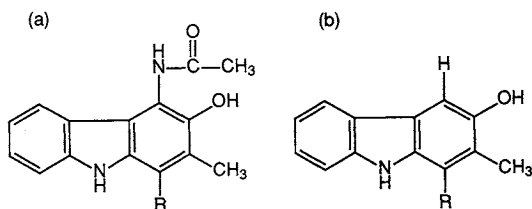


Fig. 2. Structures of antiostatins A series and carazostatin.

(a) Antiostatin series: A_1 , $\text{R}=(\text{CH}_2)_4\text{CH}_3$; A_2 , $\text{R}=(\text{CH}_2)_2\text{CH}(\text{CH}_3)\text{CH}_2\text{CH}_3$; A_3 , $\text{R}=(\text{CH}_2)_4\text{CH}(\text{CH}_3)_2$; A_4 , $\text{R}=(\text{CH}_2)_6\text{CH}_3$. (b) Carazostatin, $\text{R}=(\text{CH}_2)_6\text{CH}_3$.



two carbons and C-3 (144.3 ppm), which in turn was coupled to the hydroxy proton and amide proton. Additional ^{13}C - ^1H couplings observed between the amide carbonyl carbon (C-1'') and acetyl methyl and amide protons together with the data just described corroborated the relationship among C-1 to C-4 of the carbazole nucleus as well as the substituents on these carbons as shown in Fig. 1. Thus, the structure of antiostatin A_1 has been determined to be 1-*n*-pentyl-2-methyl-3-hydroxy-4-acetylamino-carbazole (Fig. 2).

The ^1H and ^{13}C NMR spectral comparison of antiostatin B_4 and antiostatin A_1 showed that the acetyl group in A_1 was substituted by an isobutylamino group ($-\text{NH}-\text{CH}_2$, 3.19 ppm, CH, 1.88 ppm, $2 \times \text{CH}_3$, 0.98 ppm) in B_4 in addition to the replacement of the alkyl side chain in A_1 by an *n*-heptyl residue (CH_3 , t, 0.89 ppm, $5 \times \text{CH}_2$, m, 1.28, 1.30, 1.36, 1.47 and 1.66 ppm, CH_2 , t, 2.98 ppm) in B_4 (Fig. 3).

In the HMBC spectrum of antiostatin B_4 , the methylene proton (1'-H, 2.98 ppm) of the *n*-heptyl chain was coupled to C-1 (123.7 ppm), C-2 (126.0 ppm) and C-9a (135.3 ppm). The aminomethylene proton of the isobutylamino group (3''-H, 3.19 ppm) was coupled to a carbonyl carbon

Fig. 3. Partial structure of antiostatin B₄ as revealed by HMBC analysis.

Arrows indicate ¹³C-¹H long range couplings.

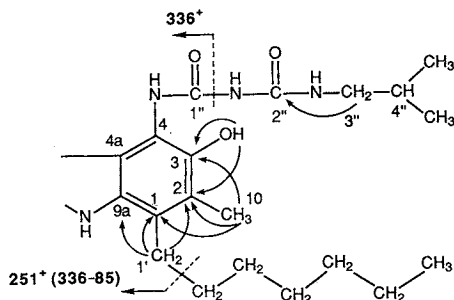
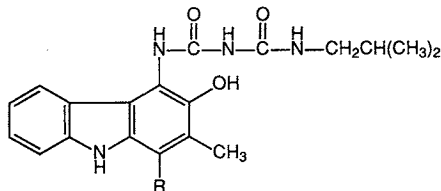


Fig. 4. Structures of antiostatin B series.

B₂, R = (CH₂)₅CH₃; B₃, R = (CH₂)₄CH(CH₃)₂;
B₄, R = (CH₂)₆CH₃; B₅, R = (CH₂)₅CH(CH₃)₂.



(C-2'', 156.8 ppm) indicating the presence of a unit (CH₃)₂CH-CH₂-NH-C(=O). Thus, there remained one carbonyl or amidino carbon (155.9 ppm), two exchangeable protons (6.92 and 10.90 ppm), one oxygen and two nitrogens to be explained. Since the carbon chemical shifts of C-4 in B₄ and A₁ were almost identical (117.6 vs. 118.3 ppm), C-4 was substituted by a nitrogen atom. Therefore, the functional group at C-4 in B₄ was either C₄-NH-C(=O)-NH-C(=O)-NH-C₄H₉ or C₄-NH-C(=NH)-O-C(=O)-NH-C₄H₉. This final problem was solved by the fragmentation ions observed in the HRFAB-MS at 336.1846 (C₂₁H₂₄N₂O₂, calcd 336.1838) and 251.0826 (C₂₁H₂₄N₂O₂-C₆H₁₃, calcd 251.0821) as shown in Fig. 3. Based on these results, the structure of antiostatin B₄ has been determined as shown in Fig. 4.

Because of the paucity of the samples available, the structures of the remaining components, antiostatins A₂, A₃, A₄ (Fig. 2), B₂, B₃ and B₅ (Fig. 4) were established by comparison of ¹H NMR and MS spectral data. Their molecular formulae and mp's are as follows; A₂ (yield 0.8 mg) C₂₁H₂₆N₂O₂, 195~197°C; A₃ (0.75 mg) C₂₂H₂₈N₂O₂, 190~192°C; A₄ (1.5 mg) C₂₂H₂₈N₂O₂, 191~193°C; B₂ (0.5 mg) C₂₅H₃₄N₄O₃, 119~120°C; B₃ (1.3

mg) C₂₆H₃₆N₄O₃, 117~118°C; B₅ (1.2 mg) C₂₇H₃₈N₄O₃, 92~94°C.

Antioostatins are the first carbazole derivatives that possess an acetamide group or a substituted urea chain in addition to a long alkyl chain. The IC₅₀ values of antiostatin A₁ and antiostatin B₄ for the assay system mentioned above were 0.207 and 0.211 μg/ml, respectively, while that of vitamin E was 10.8 μg/ml. The remaining members of the antiostatins showed similar IC₅₀ values to those of A₁ and B₄.

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